

# LERE-QSAR Analysis of Binding of $\gamma$ -Lactum Hydroxamic Acid Derivatives with Tumor Necrosis Factor-Alpha Converting Enzyme

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**Keywords:** LERE, ONIOM, Tumor necrosis factor-alpha converting enzyme

Tumor necrosis factor-alpha converting enzyme (TACE) is a metal enzyme that contains a zinc atom, and TACE converts TNF-alpha into an activated form by hydrolysis. Rheumatism arthritis and Crohn's disease are caused by overproduction of the activated form TNF-alpha. In this study, we examined the atomic and electronic mechanism underlying binding between TACE and hydroxamic acid derivatives, which have a  $\gamma$ -lactum ring [1] (Figure 1), using the ONIOM calculation and LERE (Linear Expression by Representative Energy terms)-QSAR procedure [2]. We constructed complex structures of TACE with each hydroxamic acid derivative using the MM/MD and ONIOM calculations. In the current study, we assumed that the observed overall free-energy change ( $\Delta G_{\text{obs}}$ ) in the complex formation can be expressed as a sum of the intrinsic interaction energy ( $\Delta E_{\text{bind}}^{\text{ONIOM}}$ ) and polar contribution of the solvation free-energy change ( $\Delta G_{\text{sol}}^{\text{pol}}$ ), and dispersion interaction energy change ( $E_{\text{disp}}$ ), calculated with the ONIOM(HF/6-31G: Amber) mechanical embedding (ME) method, PB calculation, and MM calculation, respectively. The sum of these three representative energy terms is nicely linear with  $\Delta G_{\text{obs}}$ , resulting in the following LERE-QSAR equation;

$$\Delta G_{\text{obs}} = 0.129 [\Delta E_{\text{bind}}(\text{ONIOM}/\text{HF}/\text{ME}) + E_{\text{disp}} + \Delta G_{\text{sol}}^{\text{pol}}] + 16.1$$

$n = 11, r = 0.912, s = 0.844 \text{ kcal/mol}$

Figure 2 shows that there is a negative correlation between  $\Delta E_{\text{bind}}(\text{ONIOM}/\text{HF}/\text{ME})$  and  $\Delta G_{\text{sol}}^{\text{pol}}$ , and that  $\Delta G_{\text{obs}}$  parallels with  $E_{\text{disp}}$ . These results suggest that the variation of  $\Delta G_{\text{obs}}$  among the derivatives is governed by  $E_{\text{disp}}$ . We also discuss a detailed binding mechanism by decomposing the binding energy obtained with the FMO calculation into individual contributions of amino acid residues.

[1] J. J. W. Duan, *et al.*, *J. Med. Chem.* **2002**, *45*, 4954–4957.

[2] T. Yoshida, S. Hitaoka, A. Mashima, T. Sugimoto, H. Matoba, and H. Chuman, *J. Phys. Chem. B*, **2012**, *116*, 10283–10289.

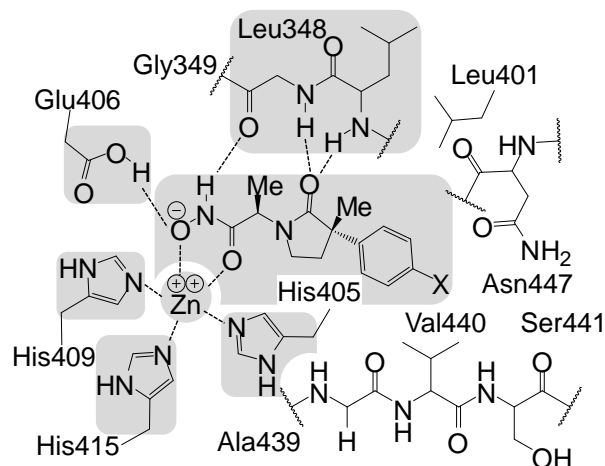


Figure 1. Schematic representation of a  $\gamma$ -lactum hydroxamic acid derivative bound in the TACE active site. Atoms in shadow areas are treated as the quantum region in the ONIOM calculation.

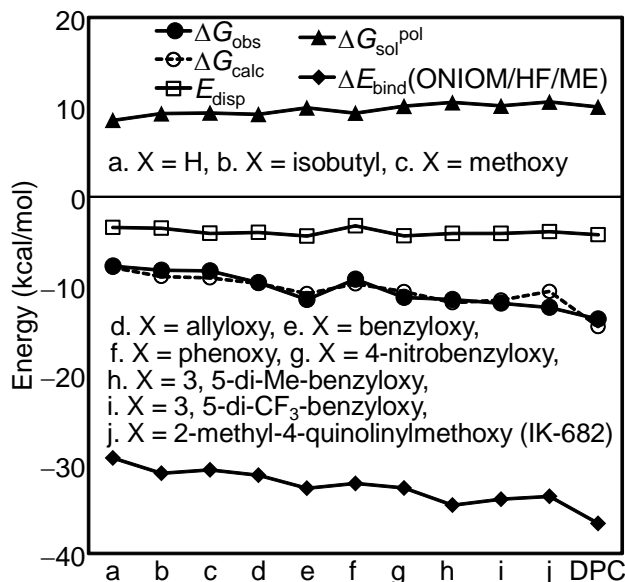


Figure 2. Variation in  $\Delta G_{\text{obs}}$ ,  $\Delta G_{\text{calc}}$ ,  $E_{\text{disp}}$ ,  $\Delta G_{\text{sol}}^{\text{pol}}$ , and  $\Delta E_{\text{bind}}(\text{ONIOM}/\text{HF}/\text{ME})$ .